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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/606,959	06/27/2003	David F. Nicoli	6000	4840
75	590 11/16/2004		EXAMINER	
Milton M. Field 2212 White Oaks Drive			WALLENHORST, MAUREEN	
Alexandria, VA			ART UNIT PAPER NUMBER 1743	
			DATE MAIL ED. 11/16/2004	

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)	
	10/606,959	NICOLI ET AL.	
Office Action Summary	Examiner	Art Unit	
	Maureen M. Wallenhorst	1743	
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence ad	ldress
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication.  - If the period for reply specified above is less than thirty (30) days, a reply If NO period for reply is specified above, the maximum statutory period w  - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, however, may a reply be time within the statutory minimum of thirty (30) days will apply and will expire SIX (6) MONTHS from cause the application to become ABANDONEI	nely filed s will be considered timel the mailing date of this c D (35 U.S.C. § 133).	
Status			
Responsive to communication(s) filed on      This action is FINAL. 2b)⊠ This      Since this application is in condition for allowar closed in accordance with the practice under E	- action is non-final. ace except for formal matters, pro		e merits is
Disposition of Claims		-	
4) ☐ Claim(s) 1-77 is/are pending in the application. 4a) Of the above claim(s) is/are withdray  5) ☐ Claim(s) is/are allowed.  6) ☐ Claim(s) 1-77 is/are rejected.  7) ☐ Claim(s) is/are objected to.  8) ☐ Claim(s) are subject to restriction and/or	vn from consideration.		
Application Papers			
9) The specification is objected to by the Examine 10) The drawing(s) filed on is/are: a) access Applicant may not request that any objection to the confidence Replacement drawing sheet(s) including the correction 11) The oath or declaration is objected to by the Examine 10.	epted or b) objected to by the Edrawing(s) be held in abeyance. See on is required if the drawing(s) is obj	e 37 CFR 1.85(a). ected to. See 37 CF	( )
Priority under 35 U.S.C. § 119	•		
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of:  1. Certified copies of the priority documents 2. Certified copies of the priority documents 3. Copies of the certified copies of the priori	s have been received. s have been received in Application ity documents have been receive (PCT Rule 17.2(a)).	on No d in this National	Stage
Attachment(s)  Notice of References Cited (PTO-892)  Notice of Draftsperson's Patent Drawing Review (PTO-948)  Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date 9/24/03.	4) Interview Summary ( Paper No(s)/Mail Da 5) Notice of Informal Pa 6) Other:	te	)-152)
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1. Claims 1-77 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

On line 6 of claim 1, the phrase "the height of said inter-particle potential energy barrier" lacks antecedent basis. See this same problem on line 4 of claims 14 and 33, on lines 6-7 of claim 40, and on lines 7-8 of claim 47. On line 5 of claim 1, it is suggested to insert the phrase – the steps of—after the word "comprising" for further clarification of the method. This same change should also be made on line 5 of claim 40.

Claim 4 is indefinite since it depends from claim 3 and recites the same thing as claim 3, i.e. that the detector is a single particle optical sensor. It is not clear how claim 4 further limits claim 3. See this same problem in claims 49-50.

On line 1 of claim 10, the phrase "the surface charge" lacks antecedent basis. See this same problem in claims 29 and 59.

On lines 2-3 of claim 13, the phrase "the oppositely charged surfaces of said particles" lacks antecedent basis. See this same problem on line 3 of claims 32 and 62. On line 3 of claims 13, the phrase "the net charge" lacks antecedent basis. See this same problem on line 3 of claims 14, 32, 33 and 62-63.

On lines 2-3 of claim 15, the phrase "the charged particles" lacks antecedent basis. See this same problem on lines 2-3 of claims 34 and 64.

In claim 22, the full meanings for the abbreviations "PDS" and "PDP" should be recited. On line 3 of claim 22, the phrase "the percentage of the dispersed phase" lacks antecedent basis. See this same problem on line 13 of claim 40, and on line 6 of claim 71.

In claim 23, the phrase "the rate of change of said PDP" lacks antecedent basis. See this same problem in claims 41, 45 and 72.

In claim 26, the full meaning for the abbreviation "FM" should be recited. This same change should also be made in claims 28, 44, 46, 75 and 77.

In claim 37, the phrase "said steps of measuring said PSD" lacks antecedent basis since claim 37 depends from claim 19, and claim 19 depends from claim 1, which does not positively recite a step of measuring PSD. See this same problem in claim 38.

On line 1 of claim 47, the phrase "The apparatus" should be changed to -An apparatus—since claim 47 is an independent claim.

In claim 56, the phrase "said high sensitivity detector" lacks antecedent basis since claim 56 depends from claim 54, and claim 54 depends from claim 47, that does not positively recite a high sensitivity detector.

- 2. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

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(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

4. Claims 1, 8, 47 and 54 are rejected under 35 U.S.C. 102(b) as being anticipated by Garver et al. (US Patent no. 6,263,725, submitted in the Information Disclosure Statement filed on September 24, 2003)

Garver et al teach of a method and apparatus for characterizing the stability of colloids (i.e. dispersions of solid or liquid particles suspended in a liquid carrier) by applying a stress factor to a sample of a colloid, and measuring the light attenuation or scattering of the sample at two or more wavelengths. The stress factor applied to the colloid is a change of temperature. The light attenuation or light scattering is measured at two or more temperatures to provide a measure of the stability of the dispersion with respect to temperature. The amount of certain colloids that are unstable to temperature variation is determined from the magnitude of the measurement change. It is inherent in the method taught by Garver et al that the instability of the colloid sample analyzed is manifested by particle agglomeration due to the reduction in the height of the potential energy barrier between the particles since Garver et al teach that the particles in the colloid sample undergo phase transition due to the change in the temperature applied, and the measurements of light attenuation and light scattering serve to give an overall particle distribution of colloidal components (i.e. agglomerated particles) therein. See the abstract, lines 55-66 in column 8 and lines 32-35 in column 9 of Garver et al.

5. Claims 1-10, 13, 47-59 and 62 are rejected under 35 U.S.C. 102(b) as being anticipated by Nicoli et al (from <u>American Laboratory</u>, vol. 33(1), January 2001, pages 32-39).

Nicoli et al teach of a method and apparatus for measuring the stability of colloidal suspensions (dispersions of solid or liquid particles suspended in a liquid carrier). Nicoli et al

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teach that the stability of colloidal suspensions can be analyzed by applying a stress factor to the suspension, such as the addition of excessive amounts of electrolyte to the suspension or by thermal or pH shock (i.e. changing the temperature or pH of the suspension). See the last column on page 34 and the first column on page 39 of Nicoli et al. Nicoli et al teach that these stress factors cause particles in the suspensions to coalesce or agglomerate due to the reduction in the inter-particle electrostatic repulsive forces that confer stability to the emulsion. See the middle column on page 34 of Nicoli et al. Nicoli et al teach that the particle size distribution of a colloidal suspension can be measured by analyzing the suspension with a single particle optical sensor (SPOS), which serves to measure both the light diffraction and light scattering of the sample. The technique of single particle optical sensing provides sensitivity to small changes in the large-particle fraction of a colloidal suspension. This sensitivity is required to ascertain the quality and stability of a product. The SPOS technique analyzes particles one at a time to produce a particle size distribution that is constructed directly from each particle. Each particle traverses a flow channel through a ribbon of light. The sensor responds to the particles by measuring both light scattering and light extinction. See page 36 in Nicoli et al. The particle size distribution (PSD) produced includes the concentration of particles as a function of size over a range of normal particle sizes and a distribution of outlier particles comprising the uppermost tail of the PSD. The outlier particles in the tail of the PSD consist of oversized fat globules caused by coalescence of the smaller primary droplets due to reduction of the inter-droplet electrostatic repulsive forces that confer stability to the emulsion.

6. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

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(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

- 7. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
  - 1. Determining the scope and contents of the prior art.
  - 2. Ascertaining the differences between the prior art and the claims at issue.
  - 3. Resolving the level of ordinary skill in the pertinent art.
  - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 8. Claims 11-12, 14, 18-22, 29-33, 37-40, 60-61, 63 and 67-71 are rejected under 35 U.S.C. 103(a) as being unpatentable over Nicoli et al (from American Laboratory). For a teaching of Nicoli et al, see previous paragraphs in this Office action.

Nicoli et al fail to teach that changing the pH level or adding electrolyte to a colloidal suspension serves to change or reduce the net charge on the surfaces of the particles in the suspension. However, such a phenomenon would have been obvious to one of ordinary skill in the art at the time of the instant invention since Nicoli et al teach that the stress factors such as pH shock and addition of electrolyte to a colloidal suspension serve to change and reduce the electrostatic repulsion between the particles, thus indicating a change or reduction of the net charge on the surfaces of the particles. Nicoli et al also fail to teach of applying the stress factors to a colloidal suspension in increments at spaced time intervals resulting in increasingly higher stress levels, or of applying different levels of stress factor to different batches of the same colloidal suspension. However, these steps would have been obvious to one of ordinary skill in the art in order to analyze and determine what level of stress factor is required to cause a

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colloidal suspension to become unstable, and how long/what level of the stress factor must be applied to the colloidal suspension in order to render it unstable. Nicoli et al also fail to teach of calculating the percentage of the dispersed phase (PDP) in the colloidal suspension analyzed from the measured particle size distribution. However, such a step would have been obvious to one of ordinary skill in the art in order to obtain a measure of the stability of the suspension since the greater the percentage of dispersed particles in the suspension, the greater the stability and vice-versa.

9. Claims 15-17, 34-36 and 64-66 are rejected under 35 U.S.C. 103(a) as being unpatentable over Nicoli et al (from American Laboratory) in view of Friberg et al (submitted in the Information Disclosure Statement filed on September 24, 2003). For a teaching of Nicoli et al, see previous paragraphs in this Office action. Nicoli et al fail to teach that that addition of a salt to a colloidal suspension can be used as a stress factor to measure the stability of the suspension.

Friberg et al teach that the addition of an electrolyte or salt such as sodium chloride to an emulsion causes a reduction in the electric repulsion potential of the particles in the emulsion, and a resulting reduction in the barrier height between the particles. This change with salt concentration results in a loss of stability in the emulsion due to the agglomeration of the particles therein. Friberg et al teach that the change from a stable emulsion to an unstable one takes place at a well-defined salt concentration. See the last half of page 66 in Friberg et al.

Based upon the combination of Nicoli et al and Friberg et al, it would have been obvious to one of ordinary skill in the art at the time of the instant invention to perform the stability testing method taught by Nicoli et al by adding a salt such as sodium chloride to a colloidal suspension as the stress factor since Friberg et al teach that the addition of a salt to an emulsion

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serves to render the emulsion unstable by reducing the repulsion potential and the barrier height between the particles, similar to the action of changing the pH or adding an electrolyte to the colloidal suspension.

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- 10. Claims 23-28, 41-46 and 72-77 would be allowable if rewritten to overcome the rejection(s) under 35 U.S.C. 112, second paragraph, set forth in this Office action and to include all of the limitations of the base claim and any intervening claims since none of the prior art of record teaches or fairly suggests a method and apparatus for measuring the stability of a colloidal suspension by applying a stress factor to the suspension, measuring the level of particle agglomeration in the suspension, and calculating a rate of change or an increase of a percentage of the dispersed phase (PDP) in the suspension along with a figure of merit (FM) derived from the calculated rate of change or increase in the PDP.
- 11. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.

Please make note of: Wells et al and Nicoli et al (US Patent no. 6,794,671) who teach of single-particle optical sensors for particle counting and sizing; and Kuhls et al and Gerst et al who teach that salts/electrolytes and pH changes can be used to render emulsions unstable (see lines 1-17 in column 5 of Kuhls et al and lines 36-45 in column 8 of Gerst et al).

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examiner should be directed to Maureen M. Wallenhorst whose telephone number is 571-272-

Any inquiry concerning this communication or earlier communications from the

1266. The examiner can normally be reached on Monday-Wednesday from 6:30 AM to 4:00

PM.

12.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Jill Warden, can be reached on 571-272-1267. The fax phone number for the

organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent

Application Information Retrieval (PAIR) system. Status information for published applications

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system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Maureen M. Wallenhorst Primary Examiner Art Unit 1743

mmw

November 15, 2004

Maureen M. Wallenhorst
PRIMARY EXAMINER
GROUP \*\*\*\* 1700